



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁵ : A61K 31/35, 9/48, 9/66	A1	(11) International Publication Number: WO 91/14429 (43) International Publication Date: 3 October 1991 (03.10.91)
(21) International Application Number: PCT/EP91/00528 (22) International Filing Date: 19 March 1991 (19.03.91) (30) Priority data: 19793 A/90 23 March 1990 (23.03.90) IT (71) Applicant (for all designated States except US): CHIESI FARMACEUTICI S.P.A. [IT/IT]; Via Palermo, 26/A, I-43100 Parma (IT). (72) Inventors; and (75) Inventors/Applicants (for US only) : CHIESI, Paolo [IT/IT]; PAVESI, Luciana [IT/IT]; Via Palermo, 26/A, I-43100 Parma (IT). (74) Agent: MINOJA, Fabrizio; Studio Consulenza Brevettuale, Via Rossini, 8, I-20122 Milano (IT).		(81) Designated States: AT (European patent), AU, BB, BE (European patent), BF (OAPI patent), BG, BJ (OAPI patent), BR, CA, CF (OAPI patent), CG (OAPI patent), CH (European patent), CM (OAPI patent), DE (European patent), DK (European patent), ES (European patent), FI, FR (European patent), GA (OAPI patent), GB (European patent), GR (European patent), HU, IT (European patent), JP, KP, KR, LK, LU (European patent), MC, MG, ML (OAPI patent), MR (OAPI patent), MW, NL (European patent), NO, PL, RO, SD, SE (European patent), SN (OAPI patent), SU, TD (OAPI patent), TG (OAPI patent), US. Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: PHARMACEUTICAL COMPOSITIONS CONTAINING IPRIFLAVONE, PROCESS FOR THE PREPARATION THEREOF AND RELATIVE THERAPEUTIC USE		
(57) Abstract Oral pharmaceutical compositions containing ipriflavone are described that are characterised by the use of oily vehicles that promote absorption of the drug enabling the dosage to be simplified.		

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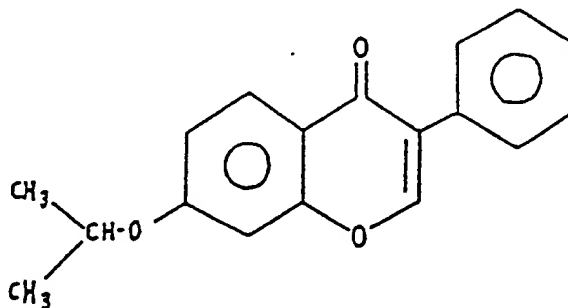
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PHARMACEUTICAL COMPOSITIONS CONTAINING IPRIFLAVONE,
PROCESS FOR THE PREPARATION THEREOF AND RELATIVE THERA-
PEUTIC USE

The present invention relates to ipriflavone oral pharmaceutical compositions and to the process for their preparation.

Ipriflavone is an isoflavone derivative of formula:



It is effective in preventing and treating postmenopausal and senile osteoporosis.

When administered orally in man, ipriflavone is absorbed rapidly and, furtherly, extensively metabolised equally rapidly: only minimal amounts of unchanged ipriflavone can be found in the blood, where its metabolites predominate. The peak plasma concentration of the unchanged drug (170 ng/ml) is reached 1.3 hours after administration, and the half-life is 9.8 hours.

Seven metabolites of ipriflavone have been identified, five of which inhibit bone resorption. The potency of metabolites 1, 2 and 5 in particular is equal to that of the precursor. It is therefore believed that the action of the antiosteoporotic properties of the

drug derive from the cumulative effect of ipriflavone and its metabolites.

The pharmaceutical composition of ipriflavone currently available is in the form of tablets each containing 200 mg of active ingredient. Bioavailability studies conducted with this formulation have shown that the optimum daily dose is 600 mg, which is achieved by administering one tablet three times daily.

Repeated drug assumptions are unpleasant to patients, especially in the case of long-term treatment where concomitant therapy is often necessary.

On the other hand, an active ingredient must be present at its site of action in adequate concentrations in order to exert its pharmacological effects. These concentrations depend on various factors such as the degree and rate of absorption, distribution and localisation in the tissues, metabolism, and elimination. All these factors are in turn affected by the chemical and physical properties of the drug molecule.

One of the basic aims of modern pharmaceutical technology is to develop appropriate release systems that will take account of all these variables and provide for each drug a dosage regimen that is more practical and acceptable to the patient and will therefore ensure more effective use of the drug in clinical practice.

The present invention relates to new oral pharmaceutical compositions of ipriflavone having the advantage of making the drug available to the body in sufficient quantities to produce the desired pharmacological response and maintain this activity for a sufficient

time for the dosage to be simplified to one or two doses daily.

A first aspect of the invention is an oral pharmaceutical composition, characterised in that the active ingredient is dispersed in an adjuvant selected from hydrogenated vegetable oils, monoglycerides, diglycerides, medium-chain triglycerides or their mixtures, white chocolate and soya lecithin, in a quantity of at least 50 % by weight per unit dose. The mixture thus obtained is enclosed in soft gelatin capsules (Scherer[®] capsules) or packed in other suitable container.

It is known that oily vehicles could be used advantageously in order to accelerate or increase the absorption of therapeutic agents with unreliable bioavailability characteristics.

However, the basic mechanisms of the interaction between drug and vehicle are still largely unknown and there are no reliable criteria that can be generally applied when developing a new formulation.

The principal parameters that determine the release characteristics of a pharmaceutical composition of this type are the affinity of the active constituent for the vehicle, the quantity and properties of the vehicle and the weight ratio of vehicle to active ingredient. It has now been found that by dispersing ipriflavone in an oily vehicle containing suitable solubilising and emulsifying agents in quantities such that the weight ratio of active principle to vehicle is at least 1:2 and enclosing this mixture in a pharmaceutical composition for oral administration an improved absorption of the drug is obtained.

A second aspect of the invention therefore relates to a process for the preparation of pharmaceutical compositions of ipriflavone for oral administration in oily vehicles.

- 5 The following examples will further illustrate the invention. The quantities of constituents given in the examples are sufficient for the preparation of 10,000 capsules.

EXAMPLE 1

- 10 Preparation of the mixture containing the active ingredient (unit dose 300 mg)

- 700 grams of hydrogenated vegetable oils are melted with heating, at a temperature not exceeding 40°C, in 2.6 kg of medium-chain triglycerides. 500 g of soya
15 lecithin are added, the whole is mixed and left to cool to room temperature. 3 kg of ipriflavone are added, and the whole is mixed for approximately 10 minutes. The mixture thus obtained is milled in a three-cylinder refiner, the paste is sieved through a 400 µ sieve, and
20 finally de-aeration is carried out.

Unit composition of the mixture containing the active ingredient.

	IPRIFLAVONE	300.0 mg
	SOYA LECITHIN	50.0 mg
25	MEDIUM-CHAIN TRIGLYCERIDES	260.0 mg
	HYDROGENATED VEGETABLE OILS	70.0 mg

	Total weight of the content	680.0 mg

- 30 The weight ratio of the constituents of the oily vehicle can be changed without any significant effect on the absorption characteristics of the active ingre-

dient.

EXAMPLE 2

Preparation of the gelatin mass for the formation of the gelatin shell.

5 Powdered gelatin, glycerol and purified water are mixed in a planetary mixer for approximately 10 minutes, after which the whole is melted in a suitable melting device for approximately 3 hours at about 70°C under vacuum. After melting, the preservatives and dyes
10 are added, selected from ethyl sodium p-hydroxybenzoate, propyl sodium p-hydroxybenzoate, red ferrous oxide, orange-yellow and titanium dioxide.

EXAMPLE 3

Preparation of the capsules.

15 The hot gelatin mass obtained as described in example 2 is used to prepare the capsules, which are filled with the mixture containing the active ingredient, using a Scherer automatic machine, by the usual known industrial methods.

20 Similarly to example 1 the following composition containing 600 mg of ipriflavone per unit dose can be prepared.

EXAMPLE 4

25 Composition containing 600 mg of ipriflavone per unit dose

Palmitic and stearic acid	56.000 mg
mono-di-tri-glycerides mixture	
White chocolate	600.000 mg
Medium-chain triglycerides	871.000 mg
30 Soya lecithin	27.000 mg
Ipriflavone	600.000 mg

Sodium saccharin	1.000 mg
Sorbitol	300.000 mg
Orange flavour	25.000 mg

5 Total weight of the content	2480.000 mg

The composition can be enclosed in a squeezeable soft gelatin capsule or in another suitable device. The container is opened at the time of administration and the content is squeezed onto a spoon or directly into
10 the oral cavity and immediately swallowed. The 600 mg ipriflavone composition is even more advantageous since it enables a once a day dosage schedule.

In-vivo bioavailability tests

The bioavailability of the pharmaceutical composition described in Examples 1-3 was determined in a
15 study conducted in 8 healthy adult volunteers between 21 and 34 years of age, in good physical condition.

The drug was administered as 300 mg capsules throughout the period of treatment (from the 1st to the
20 9th day inclusive) at the rate of one capsule twice a day (giving a total of 600 mg) taken after food at 8 am and 8 pm.

On the tenth day only the morning dose (8 am) was administered. Concomitant drug treatment was avoided.

25 Plasma and urine levels of ipriflavone and its metabolites - metabolite 1 (M1), metabolite 2 (M2), metabolite 3 (M3), metabolite 5 (M5) - were then measured.

Blood samples were collected on the first day
30 before starting the treatment, on the seventh day in the morning before meals and before drug administra-

tion, and on the tenth day at time 0 (before meals and before the planned single daily dose) and then 0.5, 1, 2, 3, 4, 8, 12, 24, 36, 48 hours after the treatment.

5 The heparinised blood samples were centrifuged within 15 minutes after collection and the separated plasma stored at -20°C until analysed by HPLC.

Urine was collected over the 24 hour period before treatment and on the tenth and eleventh days over the periods 0-24 hours and 24-48 hours after the assumption
10 of the last dose. A homogenous sample (10 ml approx.) was prepared from the total urine excreted on the tenth and eleventh days; this sample was then stored at -20°C until analysed by HPLC.

The pharmacokinetic data were compared with those
15 obtained administering an equivalent dose of ipriflavone in the standard formulation of 200 mg tablets three times daily (total dose 600 mg) according to an analogous study protocol.

Table 1 shows the main pharmacokinetic parameters
20 of ipriflavone and its metabolites after administration of 300 mg capsules according to the invention:

- area under the plasma concentration-time curve (AUC) on the 10th day at steady state during the 0-48 h period and during the dosage time interval, i.e. AUC
25 (0-8 h) for the tablets and AUC (0-12 h) for the capsules;
- maximum plasma concentration (C_{max}) calculated directly from the experimental data;
- minimum plasma concentration (C_{min}) corresponding
30 to time 0 on the 10th day (before the morning dose);
- time to reach the peak plasma level (T_{max});

- mean concentration at steady-state, calculated using the formula: $\frac{AUC}{\tau}$

where τ is the dosage interval.

5 Table 2 shows the same data for the reference composition of 200 mg standard tablets.

Table 1: Mean pharmacokinetic parameters (n=8) on ipriflavone and its metabolites M1, M2, M3, M5 after
10 administration of two 300 mg ipriflavone capsules per day (mean values \pm S.D.)

		AUC(0-48h) (ng/ml)·h	AUC(0-12h) (ng/ml)·h	C.MAX ng/ml	C.MIN ng/ml	\bar{C} ng/ml	T.MAX h
15	IPRIFLAVONE	Mean	2714.76	954.30	117.08	74.29	79.53
		S.D.	536.07	168.01	27.25	14.16	15.67
	METABOLITE 1	Mean	6038.66	2543.32	333.34	260.70	211.94
		S.D.	1298.70	553.38	56.74	52.01	46.11
20	METABOLITE 2	Mean	3082.75	1304.87	172.35	145.76	102.74
		S.D.	1216.03	519.92	65.71	62.19	43.33
	METABOLITE 3	Mean	1321.82	584.27	90.55	37.56	48.69
		S.D.	230.21	112.77	14.43	16.24	9.40
25	METABOLITE 5	Mean	9689.64	4307.15	624.92	479.68	358.93
		S.D.	1116.51	519.54	80.82	89.26	43.30

Table 2: Mean pharmacokinetic parameters (n=8) of ipriflavone and its metabolites M1, M2, M3, M5 after administration of three 200 mg tablets of ipriflavone per day (mean values + S.D.)

	AUC(0-48h) (ng/ml)*h	AUC(0-8h) (ng/ml)*h	C.MAX(a) ng/ml	C.MIN(a) ng/ml	C(a) ng/ml	T.MAX(a) h
IPRIFLAVONE	Mean	2021.57	475.42	109.90	25.13	59.43
	S.D.	1189.29	178.26	19.69	14.37	22.23
METABOLITE 1	Mean	9325.69	2369.97	397.88	346.77	269.25
	S.D.	665.86	274.46	117.53	33.03	34.31
METABOLITE 2	Mean	511.00	267.89	82.66	59.61	33.49
	S.D.	203.52	85.65	26.37	28.75	10.71
METABOLITE 3	Mean	6088.63	2058.01	369.73*	225.63*	250.22*
	S.D.	1043.12	376.64	74.78	20.54	42.34
METABOLITE 5	Mean	9021.46	3108.05	579.56	455.73	388.51
	S.D.	966.39	416.19	74.55	66.38	52.02

(a) = value compared by the Mann-Whitney U test (p < 0.05)

* = significant difference

To compare the AUC values of the two compositions at steady state on the 10th day, the areas in the respective dosage time intervals were calculated and then the AUC values of the capsules (0-12h) were multiplied by two and the AUC values of the tablets (0-8h) by three. The AUC values of the 24-hour period taking into account the different dosage scheme were so obtained.

The values of ipriflavone, the sum of the main metabolites (M1, M5) and the sum of all the metabolites (M1, M2, M3 and M5) are given in Table 3.

Table 3: Calculation of the AUC values of the total dose of 600 mg per day of ipriflavone administered over the 24-hour period (n=8)

COMPOSITION		IPRIF.	M1 + M5	M1+M2+M3+M5
<hr/>				
Capsules	Mean	1908.61	14982.87	18761.16
300 mg x 2	S.D.	376.01	2263.90	2589.26
AUC (0-12h)x2				
Tablets	Mean	1426.25	16434.06	23411.76
200 mg x 3	S.D.	534.79	1867.72	2921.44
AUC (0-8h)x3				
<hr/>				

As shown in the Tables, there were no significant difference in the pharmacokinetic behaviour of ipriflavone and its metabolites given in the two formulations.

5 The levels of metabolite 2 were higher with the capsules but this may have been due, according to our observations, to interference with the diet.

10 The plasma levels of metabolite 3 were lower with the capsules than with the tablets, but this finding is not particularly significant since this metabolite is the least important in respect of therapeutic activity.

Tables 4 and 5 show the urinary excretion values of metabolites M1, M2 and M5. There were no detectable levels of unchanged ipriflavone.

15 Over both time intervals considered, both the quantities excreted and the concentration per ml of urine found for the capsules were much higher than the respective values for the tablets.

Both these increases (total mg and mg per ml) were statistically significant for all the metabolites.

Table 4: Urinary levels of metabolite 1, metabolite 2 and metabolite 5 (10th day) after repeated administration of two 300 mg capsules of ipriflavone per day (mean values + S.D.)

		(0-24)		(24-48)	
		ml	mg	ml	mg
			%DOSE		%DOSE
METABOLITE 1	Mean	1438.63	33.37	1225.00	14.12
	S.D.	193.60	6.10	90.14	4.65
METABOLITE 2	Mean	1438.63	23.22	1225.00	8.61
	S.D.	193.60	7.91	90.14	1.76
METABOLITE, 5	Media	1438.63	109.01	1225.00	65.66
	S.D.	193.60	24.77	90.14	26.74

Table 5: Urinary levels of metabolite 1, metabolite 2 and metabolite 5 (10th day) after repeated administration of three 200 mg tablets of ipriflavone per day (mean values \pm SD.)

(0-24)				(24-48)			
	ml	mg	%DOSE	ml	mg	%DOSE	
<hr/>							
METABOLITE 1	Mean	1000.00	6.50 *	1.08	983.75	2.11 *	0.35
	S.D.	109.02	0.69	0.11	80.15	0.37	0.06
<hr/>							
METABOLITE 2	Mean	1000.00	4.21 *	0.70	983.75	1.10 *	0.18
	S.D.	109.02	0.77	0.13	80.15	0.26	0.04
<hr/>							
METABOLITE 5	Mean	1000.00	12.72 *	2.12	983.75	4.91 *	0.82
	S.D.	109.02	2.70	0.45	80.15	1.18	0.20
<hr/>							

* = significant difference compared with the capsules.

Mann-Whitney U test ($p < 0.05$)

To confirm the preliminary results we performed a further study to directly compare the bioavailability of ipriflavone and its metabolites, given orally on multiple dosing as 2 forms: 200 mg standard tablets and
5 300 mg capsules.

In this comparative study all the conditions that could influence in some measure the bioavailability behaviour such as meals, drug administration or samples collection were strictly standardized.

10 The study was carried out in twelve young healthy volunteers in good health.

All the subjects were on a standard diet for all the period of the study.

For each treatment, after an overnight fast, each
15 subject received at 8:00 a.m. on Day 1 the first oral dosing. From Day 1 to Day 10 the daily doses were administered during the meals. The two different treatments (A and B) were the following:

- Treatment A: one 200 mg Ipriflavone Tablet 3 times
20 daily at 8:00 a.m., 1:00 p.m. and 8:00 p.m.;
- Treatment B: one 300 mg Ipriflavone Scherer capsule two times daily at 8:00 a.m. and 8:00 p.m.

A wash-out period of two weeks or more was observed between two subsequent treatments.

25 Blood samples were collected on the first day before starting the treatment, and at the steady-state during day 10, following a suitable sampling schedule established in order to compare the bioavailability of the two compositions during the whole 24-hour period,
30 taking into account the different dosage scheme.

Urine samples were collected prior to the drug

administration and then on day 10, quantitatively, by fractions corresponding to time administration intervals (treatment A: 0.5/5-12/12-24 hours; treatment: B 0-12 and 12-24 hours).

5 The quantitative measurement of ipriflavone and its metabolites (M1, M2, M3 and M5) in biological specimens was performed by HPLC assay with UV detection before and after enzymatic hydrolysis of the biological samples.

10 Plasma levels of ipriflavone could not be quantified in several subjects, precluding complete pharmacokinetic study of the compound.

 This confirms the extensive metabolization of the product in man.

15 Pharmacokinetic parameters characteristic of metabolites M1, M2, M3 and M5 (free + conjugated; mean \pm sem values) are presented in Tables 6 to 9 in which C_{max}, C_{min}, AUC have the same meaning expressed before and Ae % is the total amount excreted in urine expressed
20 in percentage of the daily dose of ipriflavone administered, after correction for difference in molecular weight of the metabolites.

Table 6: Pharmacokinetic parameters characteristic of total (free + conjugated) metabolite M1, at steady-state following treatments A and B (mean \pm sem values).

	TREATMENT A (3x200mg/day)	TREATMENT B (2x300mg/day)	STATISTICS
C_{\max} (ng.ml ⁻¹)	377 \pm 59	525 \pm 58	p<0.001
AUC_{0-24} (ng.ml ⁻¹ .h)	5092 \pm 625	6651 \pm 778	p<0.01
C_{\max}/C_{\min}	3.45 \pm 0.36	3.73 \pm 0.40	NS
Ae (%)	4.84 \pm 0.51	3.41 \pm 0.41	p<0.05

Table 7: Pharmacokinetic parameters characteristic of total (free + conjugated) metabolite M2, at steady-state following treatments A and B (mean \pm sem values).

	TREATMENT A (3x200mg/day)	TREATMENT B (2x300mg/day)	STATISTICS
C_{\max} (ng.ml ⁻¹)	386 \pm 66	431 \pm 75	NS
AUC_{0-24} (ng.ml ⁻¹ .h)	5704 \pm 830	6711 \pm 1134	p<0.01
C_{\max}/C_{\min}	2.93 \pm 0.35	2.47 \pm 0.21	NS
Ae (%)	3.86 \pm 0.58	3.05 \pm 0.37	p<0.05

Table 8: Pharmacokinetic parameters characteristic of total (free + conjugated) metabolite M3, at steady-state following treatments A and B (mean \pm sem values).

	TREATMENT A (3x200mg/day)	TREATMENT B (2x300mg/day)	STATISTICS
C_{\max} (ng.ml ⁻¹)	313 \pm 44	648 \pm 96	p<0.01
AUC_{0-24} (ng.ml ⁻¹ .h)	4226 \pm 777	6701 \pm 1180	p<0.01
C_{\max}/C_{\min}	4.56 \pm 0.57	6.59 \pm 1.04	NS
Ae (%)	2.42 \pm 0.27	3.37 \pm 0.53	p<0.05

Table 9: Pharmacokinetic parameters characteristic of total (free + conjugated) metabolite M5 at steady-state following treatments A and B (mean \pm sem values).

	TREATMENT A (3x200mg/day)	TREATMENT B (2x300mg/day)	STATISTICS
C_{\max} (ng.ml ⁻¹)	747 \pm 125	1003 \pm 68	NS
AUC_{0-24} (ng.ml ⁻¹ .h)	10092 \pm 1403	12344 \pm 980	NS
C_{\max}/C_{\min}	3.39 \pm 0.43	4.20 \pm 0.37	NS
Ae (%)	12.42 \pm 1.46	13.88 \pm 1.50	NS

This study confirms that the oily vehiculation improves the absorption of ipriflavone.

Infact, after administration of capsules, metabolites M1, M2 and M3 show at steady-state, a
5 significant increase in AUC_{0-24h} . For metabolite M5, AUC_{0-24h} was not sgnificantly increased.

Ipriflavone 300 mg, administered as Scherer capsule, produces a mean increment in bioavailability equal to 35%.

10 In urine, no unchanged ipriflavone levels were found in each treatment. The urinay excretion of ipriflavone metabolites was similar (M2, M3, M5).

The simplified dosage scheme obtained with 300 mg capsules does not modify, at steady-state, the mean
15 plasma levels of ipriflavone metabolites, as demonstrated the C_{max}/C_{min} ratios calculated on a daily basis for each formulation.

The good bioavailability of capsules allows simplifying the dosage scheme by reducing the daily
20 administrations (twice daily instead of 3 times daily) and improving compliance. This fact is not negligible, also taking into account the considerable mean age of patients and long-term tratment.

CLAIMS

1. Pharmaceutical compositions for oral administration containing as active ingredient ipriflavone in combination with oily vehicles and other conventional excipients.
5
2. Pharmaceutical compositions according to Claim 1 characterised in that the unit dose of active principle is between 300 and 600 mg and that the weight ratio of active principle to vehicle material is at least 1:2.
- 10 3. Pharmaceutical compositions according to Claim 1 characterised in that the vehicle is selected from hydrogenated vegetable oils, glycerides, white chocolate, soya lecithin and their mixtures.
4. Pharmaceutical compositions according to Claims 1-3 characterised in that the vehicle is present in the unit dose in a quantity of at least 50 % by weight.
15
5. Process for the preparation of pharmaceutical compositions according to Claims 1-4, characterised in that
20 a) a mixture of glycerides, hydrogenated vegetable oils, soya lecithin optionally combined with other conventional excipients is prepared;
b) ipriflavone is added in quantities such that the weight ratio of active ingredient to vehicle material
25 is at least 1:2;
c) the mixture thus obtained is filled into a delivery system.
6. Pharmaceutical compositions according to Claims 1-5 for the treatment of osteoporosis and hypercalcaemia.
- 30 7. Pharmaceutical compositions according to claim 6 in combination with other antiosteoporotic drugs.

INTERNATIONAL SEARCH REPORT

International Application No PCT/EP 91/00528

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ⁶ According to International Patent Classification (IPC) or to both National Classification and IPC IPC ⁵ : A 61 K 31/35, A 61 K 9/48, A 61 K 9/66																				
II. FIELDS SEARCHED <div style="text-align: right; font-size: small;">Minimum Documentation Searched ⁷</div> <table style="width: 100%; border: none;"> <tr> <td style="width: 30%; border: none;">Classification System</td> <td style="border: none;">Classification Symbols</td> </tr> <tr> <td style="border: none; padding: 5px;">IPC⁵</td> <td style="border: none; padding: 5px;">A 61 K</td> </tr> </table> <div style="text-align: center; font-size: x-small; margin-top: 5px;">Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁸</div>			Classification System	Classification Symbols	IPC ⁵	A 61 K														
Classification System	Classification Symbols																			
IPC ⁵	A 61 K																			
III. DOCUMENTS CONSIDERED TO BE RELEVANT ⁹ <table style="width: 100%; border: none;"> <tr> <th style="width: 10%; border: none; font-size: x-small;">Category ⁹</th> <th style="width: 70%; border: none; font-size: x-small;">Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²</th> <th style="width: 20%; border: none; font-size: x-small;">Relevant to Claim No. ¹³</th> </tr> <tr> <td style="border: none; vertical-align: top; padding: 5px;">P, X</td> <td style="border: none; padding: 5px;"> EP, A, 0368247 (TAKEDA CHEMICAL IND.) 16 May 1990 see abstract; page 3, line 49 - page 4, line 7; example 22; claims <div style="text-align: center;">---</div> </td> <td style="border: none; vertical-align: top; text-align: center; padding: 5px;">1-7</td> </tr> <tr> <td style="border: none; vertical-align: top; padding: 5px;">X</td> <td style="border: none; padding: 5px;"> EP, A, 0129667 (TAKEDA CHEMICAL IND) 2 January 1985 see abstract, page 5, lines 1-6; page 7, lines 4-13; claims <div style="text-align: center;">---</div> </td> <td style="border: none; vertical-align: top; text-align: center; padding: 5px;">1, 3</td> </tr> <tr> <td style="border: none; vertical-align: top; padding: 5px;">Y</td> <td style="border: none; padding: 5px;"> <div style="text-align: center;">---</div> </td> <td style="border: none; vertical-align: top; text-align: center; padding: 5px;">6, 7</td> </tr> <tr> <td style="border: none; vertical-align: top; padding: 5px;">Y</td> <td style="border: none; padding: 5px;"> EP, A, 0214647 (CHINOIN) 18 March 1987 see abstract; column 1, lines 1-40 <div style="text-align: center;">---</div> </td> <td style="border: none; vertical-align: top; text-align: center; padding: 5px;">6, 7</td> </tr> <tr> <td style="border: none; vertical-align: top; padding: 5px;">A</td> <td style="border: none; padding: 5px;"> FR, M, 1065 (LAROCHÉ-NAVARRON) 15 January 1962 see the whole document, in particular page 2, column 2, lines 31-42; claim 5 <div style="text-align: center;">---</div> <div style="text-align: right;">./.</div> </td> <td style="border: none; vertical-align: top; text-align: center; padding: 5px;">1-7</td> </tr> </table>			Category ⁹	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³	P, X	EP, A, 0368247 (TAKEDA CHEMICAL IND.) 16 May 1990 see abstract; page 3, line 49 - page 4, line 7; example 22; claims <div style="text-align: center;">---</div>	1-7	X	EP, A, 0129667 (TAKEDA CHEMICAL IND) 2 January 1985 see abstract, page 5, lines 1-6; page 7, lines 4-13; claims <div style="text-align: center;">---</div>	1, 3	Y	<div style="text-align: center;">---</div>	6, 7	Y	EP, A, 0214647 (CHINOIN) 18 March 1987 see abstract; column 1, lines 1-40 <div style="text-align: center;">---</div>	6, 7	A	FR, M, 1065 (LAROCHÉ-NAVARRON) 15 January 1962 see the whole document, in particular page 2, column 2, lines 31-42; claim 5 <div style="text-align: center;">---</div> <div style="text-align: right;">./.</div>	1-7
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<div style="display: flex; justify-content: space-between; font-size: x-small;"> <div style="width: 45%;"> <p>¹⁰ Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubt on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"A" document member of the same patent family</p> </div> </div>																				
IV. CERTIFICATION <table style="width: 100%; border: none;"> <tr> <td style="width: 50%; border: none; padding: 5px;"> Date of the Actual Completion of the International Search <div style="text-align: center; font-size: large;">11th June 1991</div> </td> <td style="width: 50%; border: none; padding: 5px;"> Date of Mailing of this International Search Report <div style="text-align: center; font-size: large;">08 AUG 1991</div> </td> </tr> <tr> <td style="border: none; padding: 5px;"> International Searching Authority <div style="text-align: center; font-weight: bold;">EUROPEAN PATENT OFFICE</div> </td> <td style="border: none; padding: 5px;"> Signature of Authorized Officer <div style="text-align: center;"> MISS T. TAZELAAR </div> </td> </tr> </table>			Date of the Actual Completion of the International Search <div style="text-align: center; font-size: large;">11th June 1991</div>	Date of Mailing of this International Search Report <div style="text-align: center; font-size: large;">08 AUG 1991</div>	International Searching Authority <div style="text-align: center; font-weight: bold;">EUROPEAN PATENT OFFICE</div>	Signature of Authorized Officer <div style="text-align: center;"> MISS T. TAZELAAR </div>														
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III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)		
Category *	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages	Relevant to Claim No.
A	GB, A, 1360462 (CHINOIN) 17 July 1974 see the whole document ---	1-7
A	DE, A, 2631214 (KALI-CHEMIE PHARMA) 26 January 1978 see the whole document -----	1-7

**ANNEX TO THE INTERNATIONAL SEARCH REPORT
ON INTERNATIONAL PATENT APPLICATION NO.**

EP 9100528
SA 45802

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 23/07/91
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A- 0368247	16-05-90	JP-A- 2223533 AU-A- 4443789	05-09-90 21-06-90
EP-A- 0129667	02-01-85	JP-A- 59199630 DE-A- 3415394	12-11-84 31-10-84
EP-A- 0214647	18-03-87	JP-A- 62103077 US-A- 4826963	13-05-87 02-05-89
FR-M- 1065		None	
GB-A- 1360462	17-07-74	GB-A- 1360461	17-07-74
DE-A- 2631214	26-01-78	AU-A- 2692777 BE-A- 856756 FR-A- 2365338 JP-A- 53029917 NL-A- 7707567 SE-A- 7708040 US-A- 4202888	18-01-79 12-01-78 21-04-78 20-03-78 16-01-78 13-01-78 13-05-80